



2013 Carbon Management  
Research Symposium

# Effects of Formation Heterogeneity on CO<sub>2</sub> Gas Phase Attenuation in the Shallow Subsurface During Possible Leakage from Geologic Sequestration Sites

Michael Plampin<sup>1</sup> (mplampin@mines.edu), Toshihiro Sakaki<sup>2</sup>, Tissa Illangasekare<sup>1</sup>, and Rajesh Pawar<sup>3</sup>

<sup>1</sup>) Center for Experimental Study of Subsurface Environmental Processes (CESEP), Department of Civil & Environmental Engineering, Colorado School of Mines, Golden, Colorado, USA

<sup>2</sup>) International Services and Projects Division, National Cooperative for the Disposal of Radioactive Waste, Wetingen, Switzerland

<sup>3</sup>) Earth and Environmental Sciences Division, Los Alamos National Laboratory, Los Alamos, New Mexico, USA



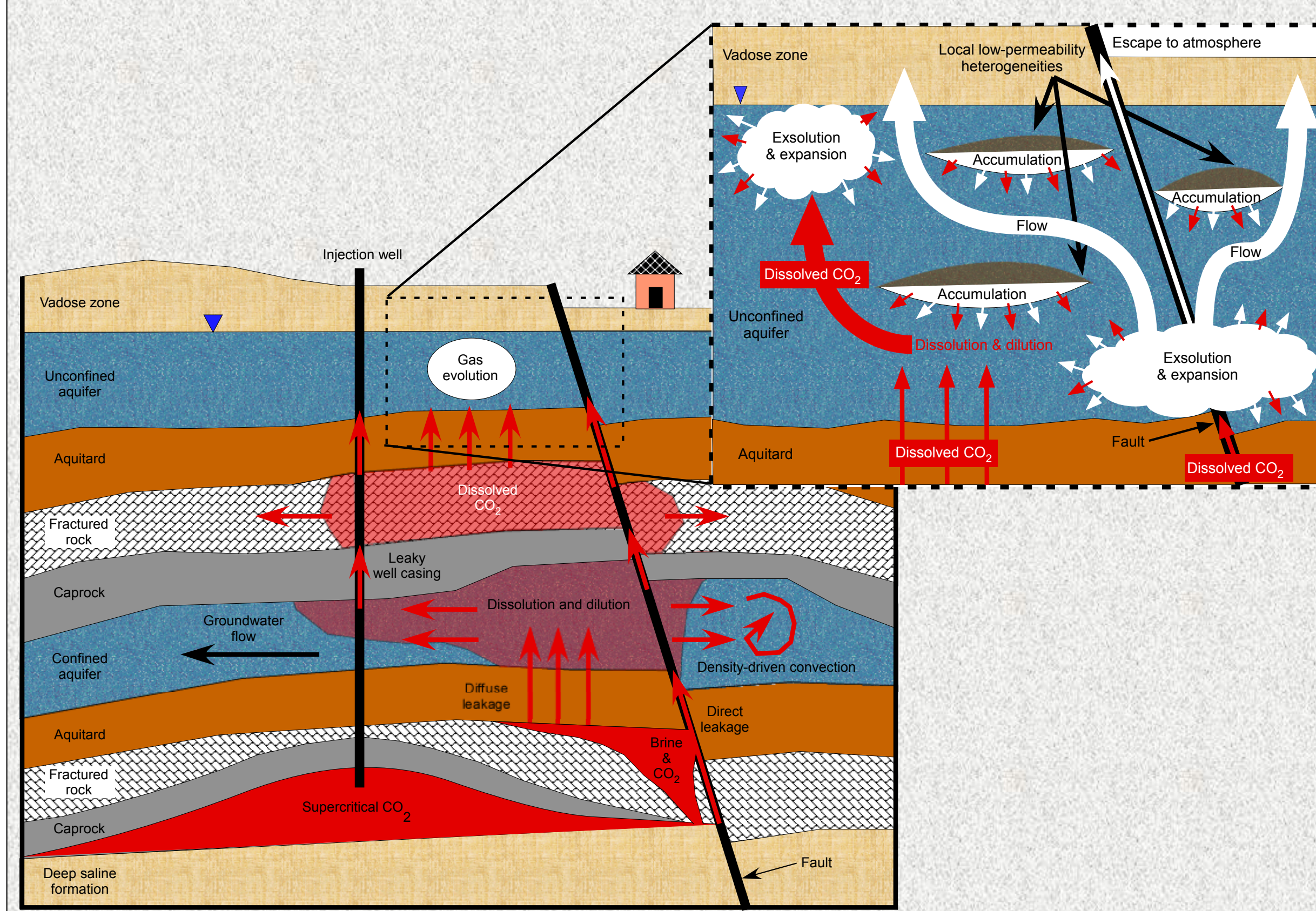
Center for Experimental Study of Subsurface  
Environmental Processes

http://cesep.mines.edu

## 1. BACKGROUND

- As a first step towards developing **risk assessment** strategies for carbon sequestration projects, we must understand **how CO<sub>2</sub> evolves** in naturally heterogeneous formations.
- As dissolved CO<sub>2</sub> migrates **upward**, water pressure decreases and gas may **exsolve**.
  - Exsolved gas partially blocks the flow of water, thus partially **attenuating** the plume of dissolved CO<sub>2</sub>.
- The spatiotemporal pattern of gas phase growth is governed by the **gravity number**, B/Ca, which quantifies the relative importance of **gravity** forces and **viscous** forces
 
$$B/Ca = \frac{\Delta\rho g k}{\mu q}$$
- Sakaki et al. (2013, in press) showed the following effects of CO<sub>2</sub> concentration on spatiotemporal gas evolution patterns using a long column with a **homogeneous** sand pack:
  - In oversaturated cases ( $P_{\text{saturation}} > P_{\text{injection}}$ ), gas formed from the **bottom up** as the CO<sub>2</sub>-water migrated up
  - In undersaturated cases ( $P_{\text{saturation}} < P_{\text{injection}}$ ), gas formed from the **top down** after the CO<sub>2</sub>-water had reached the top
- Plampin et al. (2013, in review) showed that gas phase growth can be enhanced by geologic **heterogeneity**

## 2. CONCEPTUAL MODEL



## 3. RESEARCH QUESTIONS

- How do various **soil texture interfaces** affect the formation and growth of gas phase CO<sub>2</sub> from aqueous solution in porous media?
  - Where is gas phase **first detected** with respect to the interface?
  - How fast** does the gas phase grow in each direction?
  - For what range of CO<sub>2</sub> concentrations does **heterogeneity** enhance gas phase formation (and thus CO<sub>2</sub> **attenuation**)?

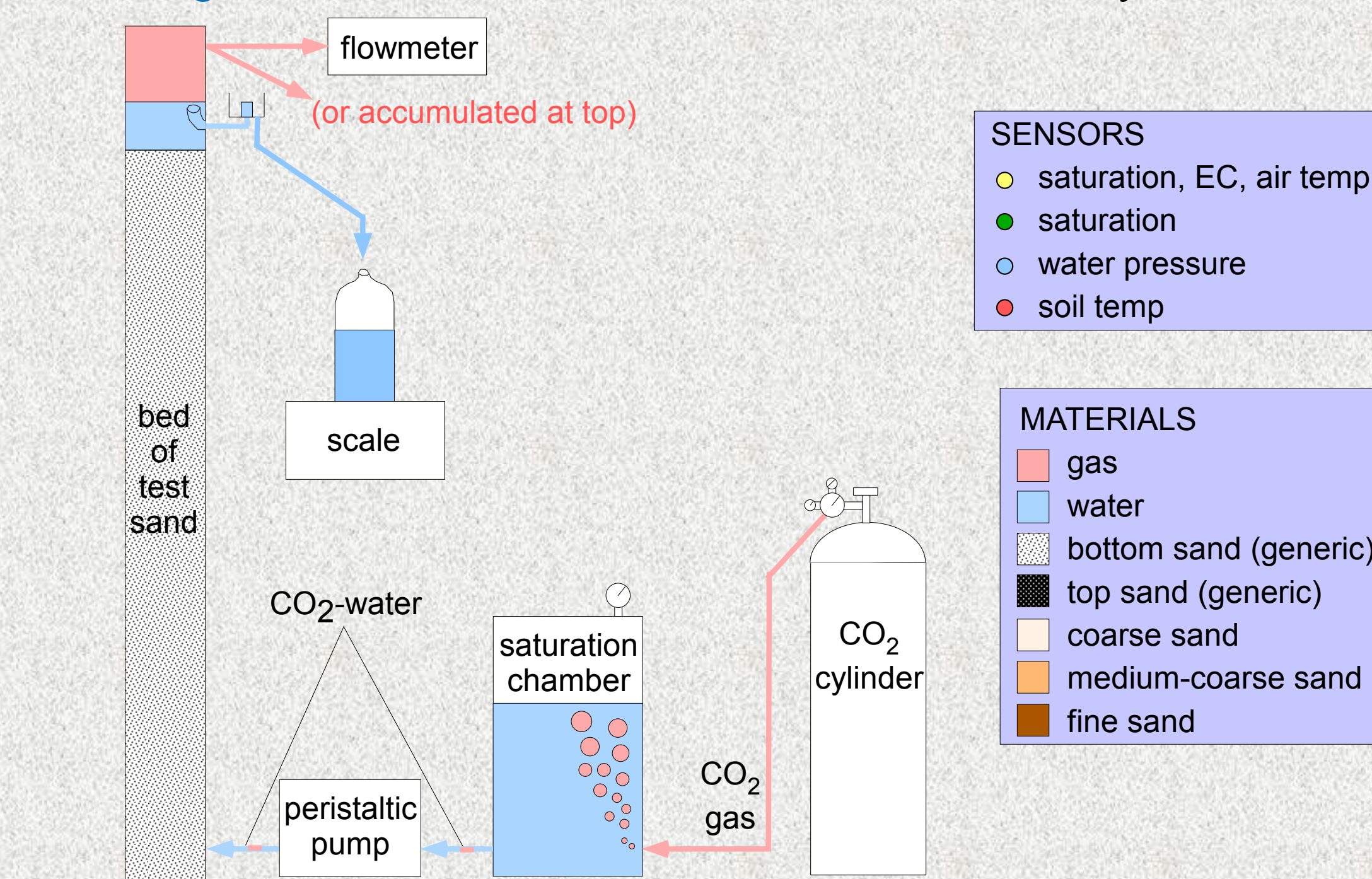
## 4. EXPERIMENTAL METHODOLOGY

### MATERIALS:

Accusand #	Description	Mean particle diameter (mm)	Intrinsic Permeability (cm <sup>2</sup> )	Porosity	Air entry pressure (kPa)	Uniformity coefficient
12/20	coarse	1.04	3.8 x 10 <sup>-6</sup>	0.32	0.8	1.23
20/30	medium-coarse	0.75	2.3 x 10 <sup>-6</sup>	0.32	1.2	1.19
50/70	fine	0.27	3.0 x 10 <sup>-7</sup>	0.34	3.4	1.20

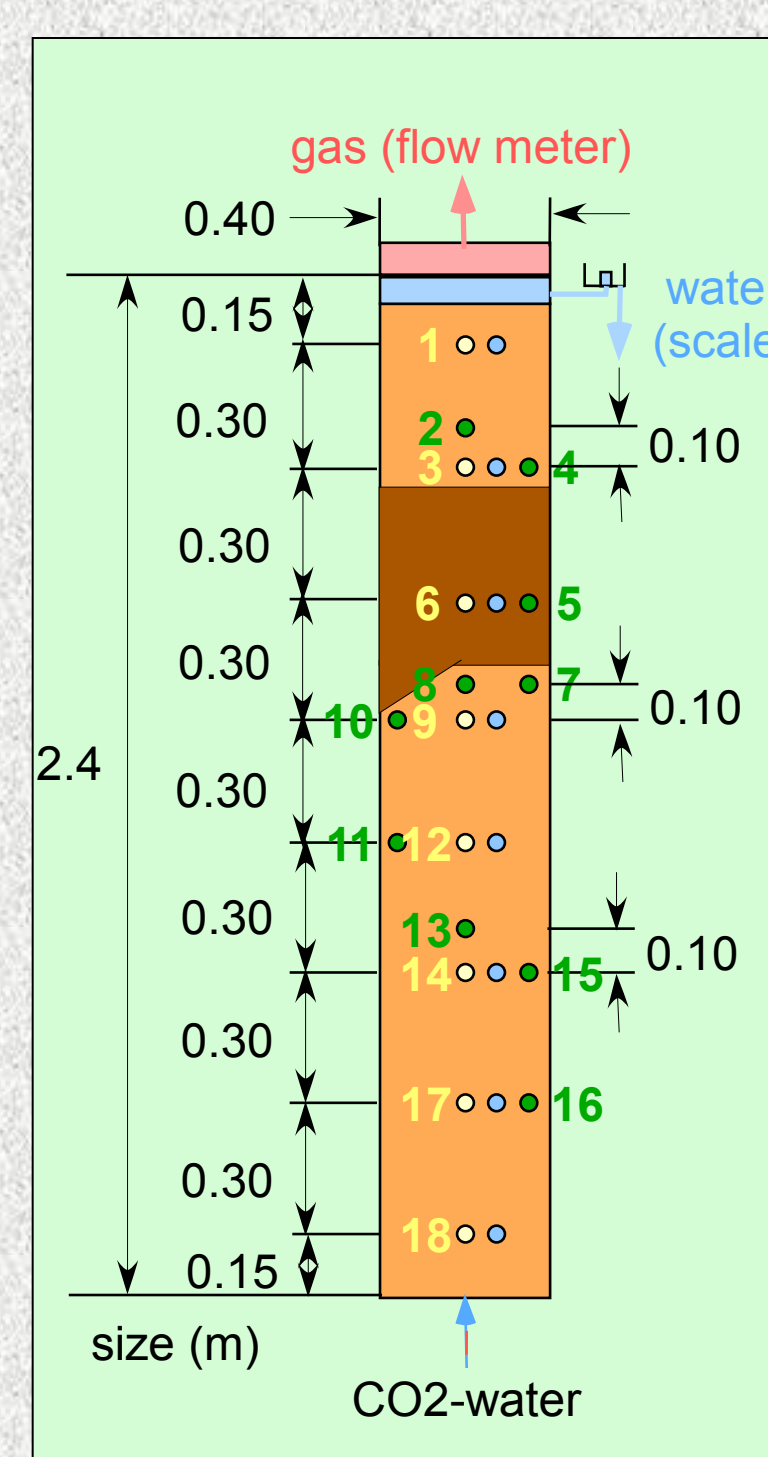
### PROCEDURE:

- Test system **wet-packed** with test sands to ensure full water saturation
- CO<sub>2</sub> gas bubbled through DI water overnight at a constant pressure (the **"saturation pressure"** or P<sub>saturation</sub>)
- CO<sub>2</sub>-saturated water **injected** through the test system from the bottom at a constant rate (the hydrostatic pressure at the injection port is P<sub>injection</sub>)
- Gas evolution behavior **monitored** through time by:
  - Soil moisture** sensors installed at various locations throughout the system
  - A **gas flow** meter attached to the outlet of the system

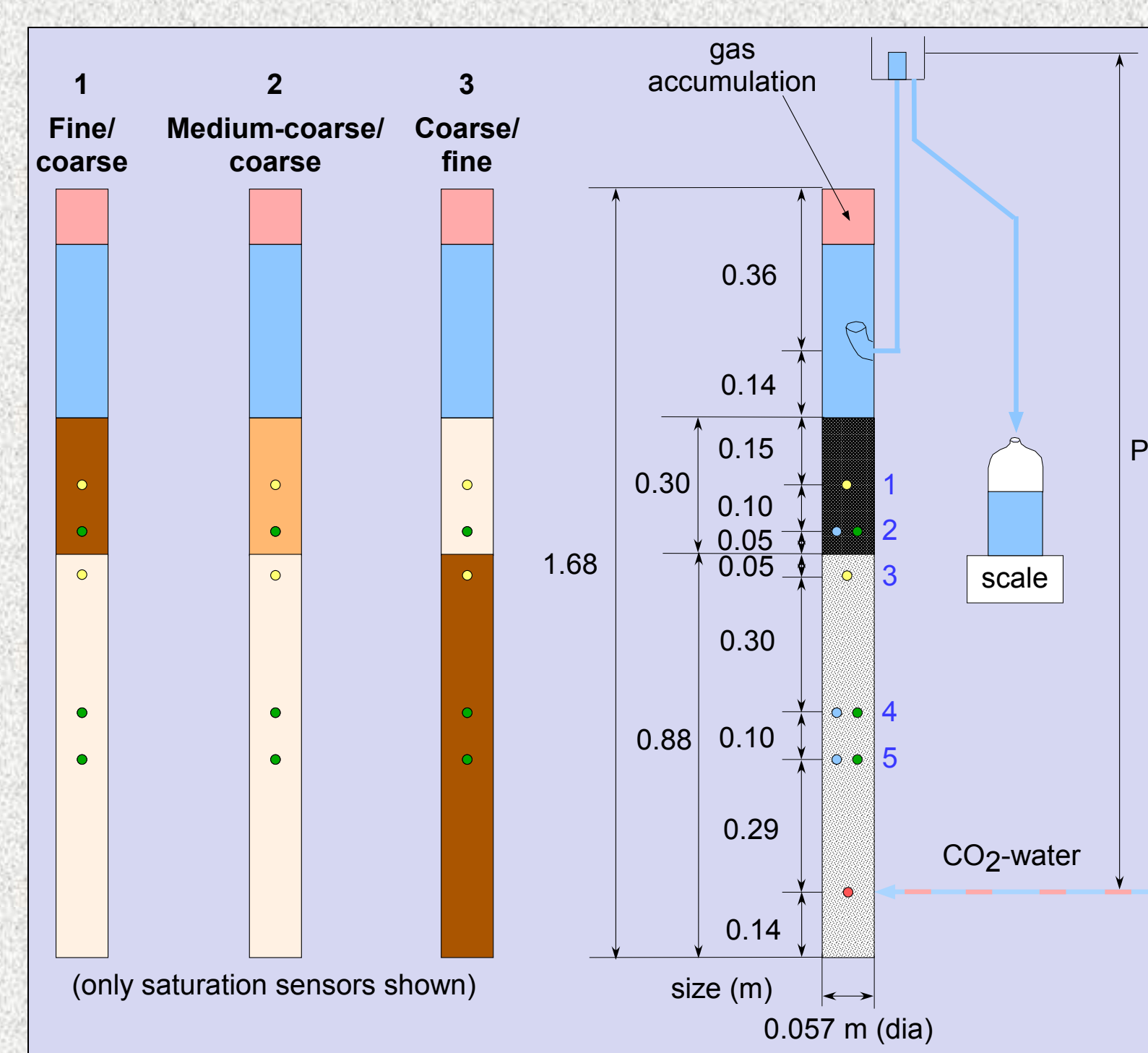


### TEST SYSTEMS & PACKING CONFIGURATIONS:

#### Rectangular Column:



#### Short Column:



## 5. EXPERIMENTAL CONDITIONS

### TERMINOLOGY:

$$\text{Oversaturation} = P_{\text{saturation}} - P_{\text{injection}}$$

where: P<sub>saturation</sub> = the constant pressure at which the injected water was **saturated** with CO<sub>2</sub> before and during the experiment, and

P<sub>injection</sub> = the **hydrostatic** pressure at the injection port

• **Oversaturated cases:** **Oversaturation > 0** (i.e., P<sub>saturation</sub> > P<sub>injection</sub>)

- Gas grows from the **bottom up** regardless of packing configuration (Sakaki et al. 2013)

• **Undersaturated cases:** **Oversaturation < 0** (i.e., P<sub>saturation</sub> < P<sub>injection</sub>)

- Gas grows from the **top down** in **homogeneous** cases (Sakaki et al. 2013)

- Heterogeneous interfaces** that exist within the **zone of oversaturation** enhance the growth of gas phase near those interfaces (Plampin et al. 2013)

➤ **Zone of oversaturation:** region where P<sub>saturation</sub> > P<sub>water</sub>

### RECTANGULAR COLUMN:

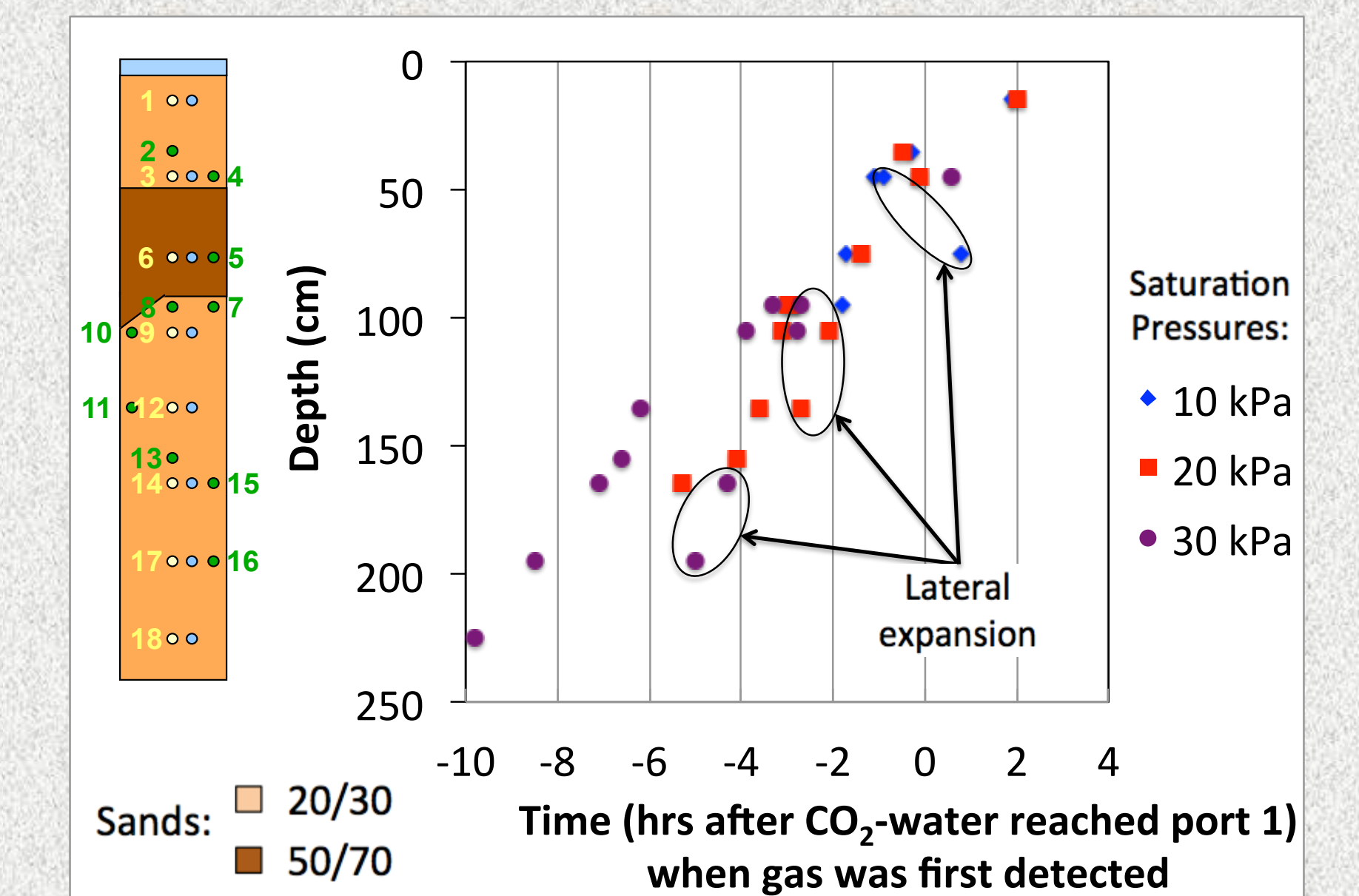
Injection rate (cm <sup>3</sup> min <sup>-1</sup> )	Saturation pressure (kPa)	B/Ca	
		Coarse sand	Fine sand
40	10		
40	20	26.5	3.5
40	30		

### SHORT COLUMN:

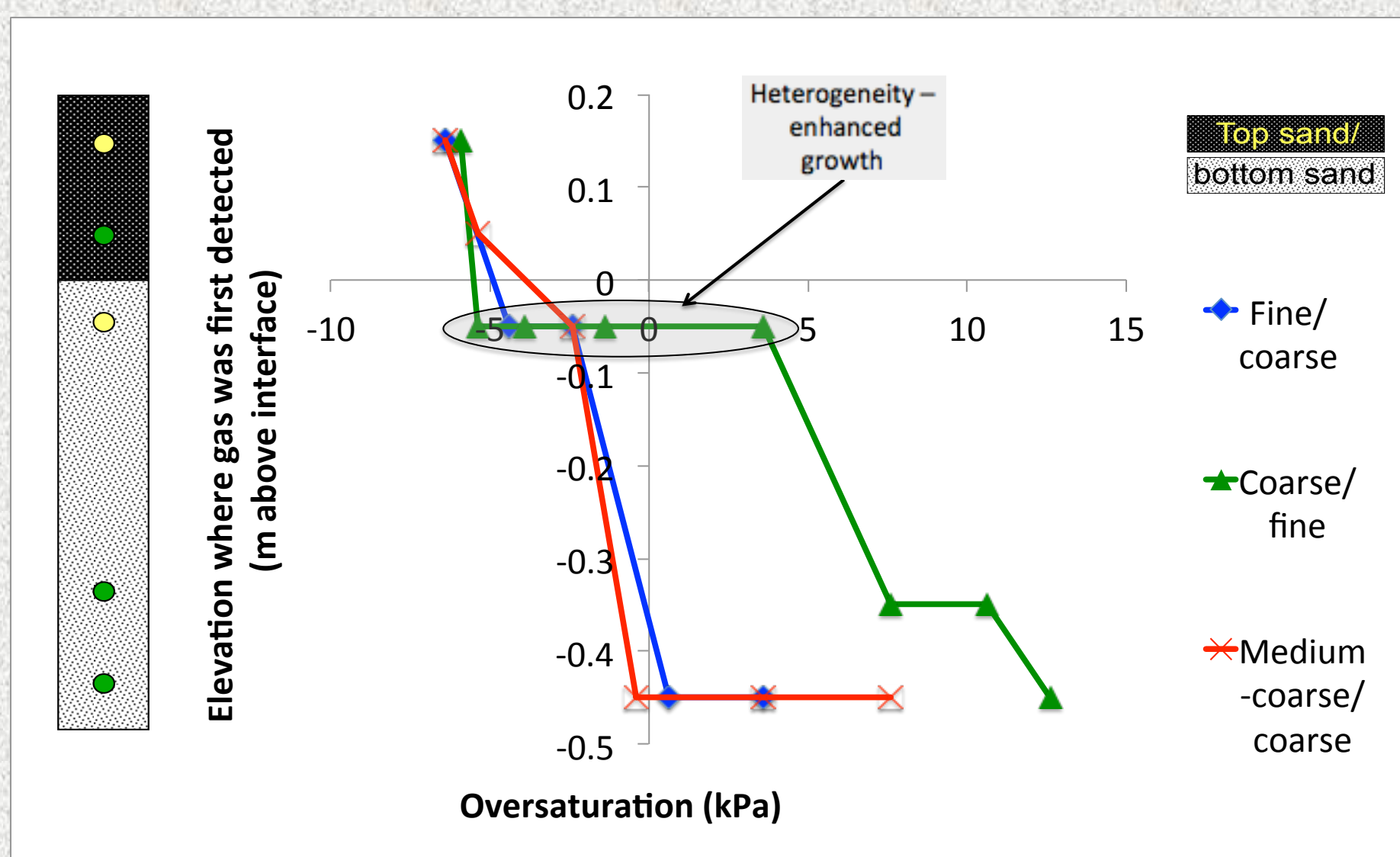
Packing configuration (top sand/ bottom sand)	Injection rate (cm <sup>3</sup> min <sup>-1</sup> )	Over-saturation (kPa)	B/Ca	
			Bottom sand	Top sand
Fine/ coarse	4	-6.4	45.7	3.8
	4	-4.4		
	4	-2.4		
	4	0.6		
Medium-coarse/ coarse	4	3.6	45.7	28.1
	4	-6.4		
	4	-5.4		
	4	-2.4		
Coarse/ fine	4	-0.4	3.8	45.7
	4	3.6		
	4	7.6		
	4	-5.9		
	4	-5.4		
	4	-3.9		
	4	-1.4		
	4	3.6		
4	12.6			

## 6. EXPERIMENTAL RESULTS

### RECTANGULAR COLUMN:



### SHORT COLUMN:



## 7. CONCLUSIONS

- Higher CO<sub>2</sub> concentrations** (i.e., higher saturation pressures) cause **faster gas phase growth** in both horizontal and vertical directions.
- There is a finite **range of oversaturation** values for which **heterogeneity controls** the rate of gas phase growth. This range is:
  - Largest for **coarse/fine** interfaces
  - Fairly small for **fine/coarse** interfaces
  - Barely present for **medium-coarse/coarse** interfaces

\* (the above descriptions are structured as follows: **"top sand/bottom sand"**)

## ACKNOWLEDGEMENTS

This research was funded partially by the US Department of Energy's Office of Fossil Energy through National Energy Technology Laboratory's CO<sub>2</sub> sequestration R&D Program (via Los Alamos National Laboratory), and partially by the National Security Science and Engineering Fellowship (NSSEFF), supported by the National Defense Education Program (NDEP).

## REFERENCES

- Sakaki, T., Plampin, M.R., Pawar, R., Komatsu, M. & Illangasekare, T.H., 2013. What controls carbon dioxide exsolution in the subsurface? ~Experimental observations in a 4.5m-long column under different heterogeneity conditions. International Journal of Greenhouse Gas Control, 13 (in press).
- Plampin, M.R., Sakaki, T., Pawar, R. & Illangasekare, T.H., 2013. Experimental Study of Gas Evolution in Heterogeneous Shallow Subsurface Formations During Leakage of Stored CO<sub>2</sub>. International Journal of Greenhouse Gas Control, 13 (in review).